

Does the concentration addition model become a more conservative predictor of aquatic metal toxicity with increasing number of metals in the mixture?

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Introduction

- The European Chemicals Strategy for Sustainability (CSS) will introduce a Mixture Allocation Factor (MAF) for Chemicals Management and other regulatory uses.
- Risk assessment of chemical mixtures is most often and most conveniently regulated following the concentration addition (CA) model.
- At low chronic effect concentrations, CA tends to overestimate mixture effects^{1,2}.

Results and discussion



- The independent action model (IA) is a better alternative for predicting chronic metal mixture toxicity^{1,2}.
- MIF (mixture interaction factor) is a quantifier used to assess the deviations of observed toxicity from toxicity predicted with CA. It indicates additivity (MIF = 1), or synergistic (MIF < 1) and antagonistic (MIF > 1) interactions, relative to CA.
- This work is part of the comprehensive Eurometaux "Metals Environmental Exposure Data" program (MEED), as project 4.

Goal

Better understanding of the joint toxicity of complex metal mixtures in realistic environmental combinations and concentrations.

 \rightarrow More accurate risk assessment of metal mixtures.

Hypotheses

- IA is generally a better predictive model than CA for chronic metal mixture toxicity.
- MIF increases with an increasing number of metals in the mixture.

Background

In a prioritization study based on metal toxicity data and European freshwater monitoring data, the metals As, Cu, Pb, Ni and Cd and the algae species Raphidocelis subcapitata were selected to test the hypotheses experimentally.

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Test design

- All 10 tests performed simultaneously
- Test 6 to 9 at equitoxic concentrations
- Test 10 at environmentally relevant concentrations
- OECD 201 guidelines 72h growth inhibition test R. subcapitata



- antagonistic effects. Ť additive synergistic Environmentally relevant ray Equitoxic ray
- All MIF higher than $1 \rightarrow CA$ overestimates mixture effects.
 - The quaternary mixture showed strong antagonism relative to CA.
 - The addition of Cd in the quinary mixture mitigated the antagonistic
 - In the quinary mixtures, the environmentally relevant ray MIF is lower than the equitoxic one.
 - No continuous increase of MIF with the number of metals in the



As + Cu + Pb + Ni + CdAs + Cu + Pb + Ni + Cdquinary (environmentally relevant ray) quinary (equitoxic ray)

Equitoxic ray \rightarrow all metals in the mixture contribute equally to the joint mixture toxicity.

Environmentally relevant ray \rightarrow concentrations and ratios based on European freshwater monitoring data to ensure environmental relevancy.

References:

1. Nys C, et al. "Systematic evaluation of chronic metal-mixture toxicity to three species and implications for risk assessment." Environmental Science & Technology 51.8 (2017): 4615-4623

2. Nys C, et al. "A framework for ecological risk assessment of metal mixtures in aquatic systems." Environmental toxicology and chemistry 37.3 (2018): 623-642

Conclusion

MIF (EC10)

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- > At environmentally relevant concentrations both CA and IA models were protective for mixture effects.
- > In most cases, IA was better in predicting mixture toxicity, while CA was the most conservative.
- \succ MIF did not increase steadily with the number of metals in the mixture.



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mixture.